This report was prepared by:
Town of Mansfield
Six Park Row
Mansfield, MA 02048
Meeting the Challenge

We are once again proud to present our annual water quality report covering all testing performed between January 1 and December 31, 2011. Over the years we have dedicated ourselves to producing drinking water that meets all state and federal standards. We continually strive to adopt new methods for delivering the best quality drinking water to you. As new challenges to drinking water safety emerge, we remain vigilant in meeting the goals of source water protection, water conservation, and community education while continuing to serve the needs of all our water users.

Please share with us your thoughts or concerns about the information in this report. After all, well-informed customers are our best allies.

For more information about this report, or for any questions relating to your drinking water, please call Kurt E. Gaffney, Water Operations Manager, at (508) 261-7376.

Community Participation

The Mansfield Board of Selectmen also serve in the roles of water and sewer commissioners for the Town. The Board meets every Wednesday evening at 7:00 p.m. at the Mansfield Town Hall (third floor, Conference Room 3A/3B), Six Park Row, Mansfield, Massachusetts. Mansfield water customers are welcome to participate in these public meetings.

Where Does My Water Come From?

The water source for the Town of Mansfield comes from the Ten Mile River Basin and the Taunton River Basin. The Town water is presently supplied from nine gravel-packed wells and one well field located in Mansfield. Cate Springs Well #1 is located off of Maple Street and pumps 1,100 gallons per minute (gpm). Albertini Well #2 pumps 500 gpm, Albertini Well #3 pumps 300 gpm, and Albertini Well #4 pumps 250 gpm; these wells are located off of West Street. Mahana Well #6 pumps 700 gpm and Morrison Well #10 pumps 695 gpm; both are located off of Plain Street. Dustin Well #7 pumps 800 gpm, Prescott Well #8 pumps 700 gpm, and Prescott Well #9 pumps 500 gpm; these wells are located off of East Street and supply a treatment facility, which is also located on East Street, to remove iron and manganese. The Walsh Wellfield, which pumps 1,042 gpm, is located off of Gilbert Street and includes a treatment facility that removes iron and manganese. A small number of residences in West Mansfield are provided water by the City of Attleboro Water System. The Town of Mansfield has interconnected and has agreements with the Towns of Easton, Norton, and Foxboro, Massachusetts to supply water in emergency situations.

What Causes the Pink Stain on Bathroom Fixtures?

The reddish-pink color frequently noted in bathrooms on shower stalls, tubs, tile, toilets, sinks, toothbrush holders, and on pets’ water bowls is caused by the growth of the bacterium Serratia marcesens. Serratia is commonly isolated from soil, water, plants, insects, and vertebrates (including man). The bacteria can be introduced into the house through any of the above-mentioned sources. The bathroom provides a perfect environment (moist and warm) for bacteria to thrive.

The best solution to this problem is to continually clean and dry the involved surfaces to keep them free from bacteria. Chlorine-based compounds work best, but keep in mind that abrasive cleaners may scratch fixtures, making them more susceptible to bacterial growth. Chlorine bleach can be used periodically to disinfect the toilet and help to eliminate the occurrence of the pink residue. Keeping bathtubs and sinks wiped down using a solution that contains chlorine will also help to minimize its occurrence.

Serratia will not survive in chlorinated drinking water.
Substances That Could Be in Water

To ensure that tap water is safe to drink, the Massachusetts Department of Environmental Protection (DEP) and the U.S. Environmental Protection Agency (U.S. EPA) prescribe regulations that limit the amount of certain contaminants in water provided by public water systems. The Food and Drug Administration (FDA) and the Massachusetts Department of Public Health (DPH) regulations establish limits for contaminants in bottled water, which must provide the same protection for public health. Drinking water, including bottled water, may reasonably be expected to contain at least small amounts of some contaminants. The presence of these contaminants does not necessarily indicate that the water poses a health risk.

The sources of drinking water (both tap water and bottled water) include rivers, lakes, streams, ponds, reservoirs, springs, and wells. As water travels over the surface of the land or through the ground, it dissolves naturally occurring minerals and, in some cases, radioactive material and can pick up substances resulting from the presence of animals or from human activity. Substances that may be present in source water include: Microbial Contaminants, such as viruses and bacteria, which may come from sewage treatment plants, septic systems, agricultural livestock operations, or wildlife; Inorganic Contaminants, such as salts and metals, which can be naturally occurring or may result from urban stormwater runoff, industrial or domestic wastewater discharges, oil and gas production, mining, or farming; Pesticides and Herbicides, which may come from a variety of sources such as agriculture, urban stormwater runoff, and residential uses; Organic Chemical Contaminants, including synthetic and volatile organic chemicals, which are by-products of industrial processes and petroleum production, and which may also come from gas stations, urban stormwater runoff, and septic systems; and Radioactive Contaminants, which can be naturally occurring or may be the result of oil and gas production and mining activities.

More information about contaminants and potential health effects can be obtained by calling the U.S. EPA's Safe Drinking Water Hotline at (800) 426-4791.

Violation Information

Mansfield routinely monitors for the presence of drinking water contaminants to ensure the safety of the water supply. On August 16 and August 18, 2011, two of six “untreated” raw ground water samples collected at Dustin Well #7 tested positive for E.coli or enterococci, which is a fecal indicator. Dustin Well #7 is one of ten wells that supplies water to the drinking water system.

During normal operations, raw water from Dustin Well #7 is pumped from the ground source and treated with chlorine to kill viruses and fecal indicators, including E.coli or enterococci. Only the chlorinated water is distributed into the public drinking water system. In 2011, zero samples collected from the chlorine-treated drinking water tested positive for E.coli or enterococci.

To comply with a new Massachusetts Drinking Water Regulations Ground Water Rule, the Town turned off Well #7, issued a public notice in the local newspaper, and implemented system improvements to achieve 4-log disinfection removal certification. After receiving 4-log certification in November 2011, the well was put back in use.

Fecal indicators are microbes whose presence indicates that the water may be contaminated with human or animal wastes. Microbes in these wastes can cause short-term health effects, such as diarrhea, cramps, nausea, headaches, or other symptoms. They may pose a special health risk for infants, young children, some of the elderly, and people with severely compromised immune systems.

Lead in Home Plumbing

If present, elevated levels of lead can cause serious health problems, especially for pregnant women and young children.Lead in drinking water is primarily from materials and components associated with service lines and home plumbing. We are responsible for providing high-quality drinking water, but we cannot control the variety of materials used in plumbing components. When your water has been sitting for several hours, you can minimize the potential for lead exposure by flushing your tap for 30 seconds to 2 minutes before using water for drinking or cooking. If you are concerned about lead in your water, you may wish to have your water tested. Information on lead in drinking water, testing methods, and steps you can take to minimize exposure is available from the Safe Drinking Water Hotline or at www.epa.gov/safewater/lead.
What's a Cross-connection?

Cross-connections that contaminate drinking water distribution lines are a major concern. A cross-connection is formed at any point where a drinking water line connects to equipment (boilers), systems containing chemicals (air conditioning systems, fire sprinkler systems, irrigation systems), or water sources of questionable quality. Cross-connection contamination can occur when the pressure in the equipment or system is greater than the pressure inside the drinking water line (backpressure). Contamination can also occur when the pressure in the drinking water line drops due to fairly routine occurrences (main breaks, heavy water demand), causing contaminants to be sucked out from the equipment and into the drinking water line (backsiphonage).

Outside water taps and garden hoses tend to be the most common sources of cross-connection contamination at home. The garden hose creates a hazard when submerged in a swimming pool or when attached to a chemical sprayer for weed killing. Garden hoses that are left lying on the ground may be contaminated by fertilizers, cesspools, or garden chemicals. Improperly installed valves in your toilet could also be a source of cross-connection contamination.

Community water supplies are continuously jeopardized by cross-connections unless appropriate valves, known as backflow prevention devices, are installed and maintained. For more information, review the Cross-Connection Control Manual from the U.S. EPA's Web site at http://water.epa.gov/infrastructure/drinkingwater/pws/crossconnectioncontrol/index.cfm. You can also call the Safe Drinking Water Hotline at (800) 426-4791.

Water Main Flushing

Distribution mains (pipes) convey water to homes, businesses, and hydrants in your neighborhood. The water entering distribution mains is of very high quality; however, water quality can deteriorate in some areas of the distribution mains over time. Water main flushing is the process of cleaning the interior of the water distribution mains by sending a rapid flow of water through the mains.

Flushing maintains water quality in several ways. For example, flushing removes sediments like iron and manganese. Although iron and manganese do not pose health concerns, they can affect the taste, clarity, and color of the water. Additionally, sediments can shield microorganisms from the disinfecting power of chlorine, contributing to the growth of microorganisms within the distribution mains. Flushing helps remove stale water and ensures the presence of fresh water with sufficient dissolved oxygen, disinfectant levels, and an acceptable taste and smell.

During flushing operations in your neighborhood, some short-term deterioration of water quality, though uncommon, is possible. You should avoid tap water for household uses at that time. If you do use the tap, allow your cold water to run for a few minutes at full velocity before use, and avoid using hot water to prevent sediment accumulation in your hot water tank.

Please contact us if you have any questions or if you would like more information on our water main flushing schedule.

Important Health Information

Some people may be more vulnerable to contaminants in drinking water than the general population. Immunocompromised persons such as persons with cancer undergoing chemotherapy, persons who have undergone organ transplants, people with HIV/AIDS or other immune system disorders, some elderly, and infants may be particularly at risk from infections. These people should seek advice about drinking water from their health care providers. The U.S. EPA/CDC (Centers for Disease Control and Prevention) guidelines on appropriate means to lessen the risk of infection by Cryptosporidium and other microbial contaminants are available from the Safe Drinking Water Hotline at (800) 426-4791 or http://water.epa.gov/drink/hotline.

The Benefits of Fluoridation

Fluoride is a naturally occurring element in many water supplies in trace amounts. In our system, the fluoride level is adjusted to an optimal level averaging one part per million (ppm) to improve oral health in children. At this level, it is safe, odorless, colorless, and tasteless. Our water system has been providing this treatment since 1997. Over 3.9 million people served by the 140 Massachusetts water systems and 184 million people in the U.S. receive the health and economic benefits of fluoridation.
Fact or Fiction

Tap water is cheaper than soda pop. *(Fact: You can refill an 8 oz. glass of tap water approximately 15,000 times for the same cost as a six-pack of soda pop. And, water has no sugar or caffeine.)*

Methods for the treatment and filtration of drinking water were developed only recently. *(Fiction: Ancient Egyptians treated water by siphoning water out of the top of huge jars after allowing the muddy water from the Nile River to settle. And, Hippocrates, known as the father of medicine, directed people in Greece to boil and strain water before drinking it.)*

A typical shower with a non-low-flow showerhead uses more water than a bath. *(Fiction: A typical shower uses less water than a bath.)*

Water freezes at 32 degrees Fahrenheit. *(Fiction: You can actually chill very pure water past its freezing point (at standard pressure) without it ever becoming solid.)*

The Pacific Ocean is the largest ocean on Earth. *(Fact: The Atlantic Ocean is the second largest and the Indian Ocean is the third largest.)*

A single tree will give off 70 gallons of water per day in evaporation. *(Fact)*
During the past year we have taken hundreds of water samples in order to determine the presence of any radioactive, biological, inorganic, volatile organic, or synthetic organic contaminants. The tables below show only those contaminants that were detected in the water. The state allows us to monitor for certain substances less often than once per year because the concentrations of these substances do not change frequently. In these cases, the most recent sample data are included, along with the year in which the sample was taken.

### REGULATED SUBSTANCES

<table>
<thead>
<tr>
<th>SUBSTANCE</th>
<th>UNIT OF MEASURE</th>
<th>YEAR SAMPLED</th>
<th>MCL [MRDL]</th>
<th>MCLG [MRDLG]</th>
<th>AMOUNT DETECTED</th>
<th>RANGE LOW-HIGH</th>
<th>VIOLATION</th>
<th>TYPICAL SOURCE</th>
</tr>
</thead>
<tbody>
<tr>
<td>Chlorine</td>
<td>(ppm)</td>
<td>2011</td>
<td>[4]</td>
<td>[4]</td>
<td>0.18</td>
<td>ND–1.03</td>
<td>No</td>
<td>Water additive used to control microbes</td>
</tr>
<tr>
<td>Combined Radium</td>
<td>(pCi/L)</td>
<td>2009</td>
<td>5</td>
<td>0</td>
<td>0.66</td>
<td>NA</td>
<td>No</td>
<td>Erosion of natural deposits</td>
</tr>
<tr>
<td>Fecal Indicators</td>
<td>[E. coli, enterococci or coliphage] Ground Water Rule (# positive samples)</td>
<td>2011</td>
<td>TT</td>
<td>NA</td>
<td>2</td>
<td>NA</td>
<td>Yes</td>
<td>Human and animal fecal waste</td>
</tr>
<tr>
<td>Fluoride</td>
<td>(ppm)</td>
<td>2011</td>
<td>4</td>
<td>4</td>
<td>0.94</td>
<td>0.81–1.08</td>
<td>No</td>
<td>Water additive that promotes strong teeth</td>
</tr>
<tr>
<td>HX Acids [HAA]</td>
<td>(ppb)</td>
<td>2011</td>
<td>60</td>
<td>NA</td>
<td>6.55</td>
<td>ND–20.6</td>
<td>No</td>
<td>By-product of drinking water disinfection</td>
</tr>
<tr>
<td>Nitrate</td>
<td>(ppm)</td>
<td>2011</td>
<td>10</td>
<td>10</td>
<td>0.53</td>
<td>ND–1.61</td>
<td>No</td>
<td>Runoff from fertilizer use; Leaching from septic tanks, sewage; Erosion of natural deposits</td>
</tr>
<tr>
<td>Perchlorate</td>
<td>(ppb)</td>
<td>2011</td>
<td>2</td>
<td>NA</td>
<td>0.21</td>
<td>0.06–0.32</td>
<td>No</td>
<td>Inorganic chemicals used as oxidizers in solid propellants for rockets, missiles, fireworks and explosives</td>
</tr>
<tr>
<td>TTHMs [Total Trihalomethanes]</td>
<td>(ppb)</td>
<td>2011</td>
<td>80</td>
<td>NA</td>
<td>26.5</td>
<td>2.6–50.3</td>
<td>No</td>
<td>By-product of drinking water disinfection</td>
</tr>
<tr>
<td>Total Coliform Bacteria</td>
<td>(# positive samples)</td>
<td>2011</td>
<td>1 positive monthly sample</td>
<td>0</td>
<td>2</td>
<td>NA</td>
<td>No</td>
<td>Naturally present in the environment</td>
</tr>
</tbody>
</table>

Tap water samples were collected for lead and copper analyses from sample sites throughout the community.

<table>
<thead>
<tr>
<th>SUBSTANCE</th>
<th>UNIT OF MEASURE</th>
<th>YEAR SAMPLED</th>
<th>AL</th>
<th>MCLG</th>
<th>AMOUNT DETECTED (90TH% TILE)</th>
<th>SITES ABOVE AL/TOTAL SITES</th>
<th>VIOLATION</th>
<th>TYPICAL SOURCE</th>
</tr>
</thead>
<tbody>
<tr>
<td>Copper</td>
<td>(ppm)</td>
<td>2010</td>
<td>1.3</td>
<td>1.3</td>
<td>0.67</td>
<td>0/30</td>
<td>No</td>
<td>Corrosion of household plumbing systems; Erosion of natural deposits</td>
</tr>
<tr>
<td>Lead</td>
<td>(ppb)</td>
<td>2010</td>
<td>15</td>
<td>0</td>
<td>4</td>
<td>0/30</td>
<td>No</td>
<td>Corrosion of household plumbing systems; Erosion of natural deposits</td>
</tr>
</tbody>
</table>

### UNREGULATED SUBSTANCES ¹

<table>
<thead>
<tr>
<th>SUBSTANCE</th>
<th>UNIT OF MEASURE</th>
<th>YEAR SAMPLED</th>
<th>AMOUNT DETECTED</th>
<th>RANGE LOW-HIGH</th>
<th>TYPICAL SOURCE</th>
</tr>
</thead>
<tbody>
<tr>
<td>Bromodichloromethane</td>
<td>(ppb)</td>
<td>2011</td>
<td>1.3</td>
<td>ND–3.9</td>
<td>By-product of drinking water disinfection</td>
</tr>
<tr>
<td>Chlorodibromomethane</td>
<td>(ppb)</td>
<td>2011</td>
<td>0.33</td>
<td>ND–1.0</td>
<td>By-product of drinking water disinfection</td>
</tr>
<tr>
<td>Chloroform</td>
<td>(ppb)</td>
<td>2011</td>
<td>2.7</td>
<td>ND–6.9</td>
<td>By-product of drinking water disinfection</td>
</tr>
<tr>
<td>Sodium</td>
<td>(ppm)</td>
<td>2011</td>
<td>31.1</td>
<td>5.65–47.4</td>
<td>Naturally occurring</td>
</tr>
</tbody>
</table>

### INITIAL DISTRIBUTION SYSTEM EVALUATION (IDSE) ²

<table>
<thead>
<tr>
<th>SUBSTANCE</th>
<th>UNIT OF MEASURE</th>
<th>YEAR SAMPLED</th>
<th>AMOUNT DETECTED</th>
<th>RANGE LOW-HIGH</th>
<th>TYPICAL SOURCE</th>
</tr>
</thead>
<tbody>
<tr>
<td>Haloacetic Acids [HAA]–IDSE Results</td>
<td>(ppb)</td>
<td>2009</td>
<td>7.31</td>
<td>ND–35.1</td>
<td>By-product of drinking water disinfection</td>
</tr>
<tr>
<td>TTHMs [Total Trihalomethanes]–IDSE Results</td>
<td>(ppb)</td>
<td>2009</td>
<td>16.20</td>
<td>ND–49.9</td>
<td>By-product of drinking water disinfection</td>
</tr>
</tbody>
</table>
Unregulated contaminants are those for which the U.S. EPA has not established drinking water standards. The purpose of unregulated contaminant monitoring is to assist the U.S. EPA in determining their occurrence in drinking water and whether future regulation is warranted.

We were required by the U.S. EPA to conduct an evaluation of our distribution system. This is known as an Initial Distribution System Evaluation (IDSE) and is intended to identify locations in our distribution system that have elevated disinfection by-product concentrations. Disinfection by-products (e.g., HAAs and TTHMs) result from continuous disinfection of drinking water. They form when disinfectants combine with organic matter that naturally occurs in the source water.

### Definitions

**90th Percentile:** Out of every 10 homes sampled, 9 were at or below this level.

**AL (Action Level):** The concentration of a contaminant which, if exceeded, triggers treatment or other requirements that a water system must follow.

**MCL (Maximum Contaminant Level):** The highest level of a contaminant that is allowed in drinking water. MCLs are set as close to the MCLGs as feasible using the best available treatment technology.

**MCLG (Maximum Contaminant Level Goal):** The level of a contaminant in drinking water below which there is no known or expected risk to health. MCLGs allow for a margin of safety.

**MRDL (Maximum Residual Disinfectant Level):** The highest level of a disinfectant allowed in drinking water. There is convincing evidence that addition of a disinfectant is necessary for control of microbial contaminants.

**MRDLG (Maximum Residual Disinfectant Level Goal):** The level of a drinking water disinfectant below which there is no known or expected risk to health. MRDLGs do not reflect the benefits of the use of disinfectants to control microbial contaminants.

**NA:** Not applicable

**ND (Not detected):** Indicates that the substance was not found by laboratory analysis.

**pCi/L (Picocuries per liter):** A measure of radioactivity.

**ppb (parts per billion):** One part substance per billion parts water (or micrograms per liter).

**ppm (parts per million):** One part substance per million parts water (or milligrams per liter).

**TT (Treatment Technique):** A required process intended to reduce the level of a contaminant in drinking water.